

O- cresol Biodegradation by *Bacillus badius* D1 isolated from Salt Lake Lonar (MS), India.

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Abstract: Cresols are known to be dominant lethal mutations in germ cells in male mice. Cresols may persist in extremely oligotrophic water bodies, and/or those under anaerobic conditions, such as in some sediments and groundwater aquifers. The Strain *Bacillus badius* D1 was employed for biodegradation of o-cresol at higher concentration isolated from Lonar Crater Lake Buldana (MS), India. It has degraded 350 mg/L concentration within 30 hr. The optimum pH and temperature observed was 9.00 and 32 °C respectively. The intermediates of catabolism were trapped as salicylic acid and catechol. This organism can be employed for bioremediation at polluted site or in bioreactors.

Key Words: *Bacillus badius*, alkaliphile, o-cresol, biodegradation

I. Introduction:

Cresols are widespread in nature, occurring in petroleum, coal tar, crude oil and volcanic actions. They are emitted from municipal incinerators, during coal and wood combustion, with vehicle exhaust, from oil refineries and cigarette smoke. Cresols are also products of the photo oxidation of toluene.

These could be absorbed through the skin, inhalation and are distributed throughout the body. These are toxic to respiratory and gastrointestinal tract. The p-cresol on oxidation leads to a reactive quinone methide. Cresols are known to be dominant lethal mutations in germ cells in male mice. Cresols may persist in extremely oligotrophic water bodies, and/or those under anaerobic conditions, such as in some sediments and groundwater aquifers [1]. m -Cresol, p-cresol o-cresol have high (21.5 - 24.4 g/L) water solubility [2] at 25°C. Hence the risk of toxicity increases in water bodies.

Biodegradation is the dominant mechanism responsible for the fast breakdown of cresols in soil and water. Although bio-degradation of cresol has reported by various researchers using neutrophiles but there is no report of o-cresol degradation by an alkaliphile. An attempt was made to study the o-cresol bio-degradation in alkaline condition.

II. 2 Methodology:

2.1 Chemicals: Yeast extract, peptone was purchased from Hi-media Mumbai. Other chemicals were taken from SRL chemicals. Pure o-cresol was obtained by the courtesy of Department of Chemistry, Pune University.

2.2 Biodegradation study:

Six 500 ml conical flasks containing sterilized 250 ml alkaline broth of pH-9.00 were inoculated by 1 % *Bacillus badius* D1 culture possessing 1.6 OD at 600 nm aseptically [3]. These culture flasks were incubated for 24 hrs at 37 °C with shaking on Orbital shaker at 110 rpm. The 24 hrs grown culture flasks were induced by adding concentration (150 -350) mg/L of o-cresol. These flasks were removed sequentially from 0 to 30 hours by 6 hr. interval. The removed flasks were used for OD at 600 nm to check the growth and then spun to DuPont Sorvall Cold centrifuge at 10000 x g. Similarly one another flask was kept as abiotic control by adding experimental concentration of cresol. The residual concentration was investigated using 4 aminoantipyrine method using Jasco Varian -630 as our early report [4,5,6]. These experiments were repeated thrice. The study was extended for various concentration, pH and temperature. The residual sample was given for GCMS analysis after solvent extraction. The study was extended to the bio-catalytic induction using standard methods like Omura and Sato[7,8] for cytochrome P450. Catechol 1, 2 dioxygenase activity was studied by Guzik Urszula, Gren Izabela et al [9] and catechol 2, 3 dioxygenase activity was studied by J.M Sala Trepat and W.C. Evans[10,11]. The enzyme activity noted as $\mu\text{M}/\text{min}/\text{mg}$ of protein. The enzyme activity was expressed as $\mu\text{M}/\text{min}/\text{mg}$ of protein.

2.3 Spectrophotometric pattern of biodegradation:

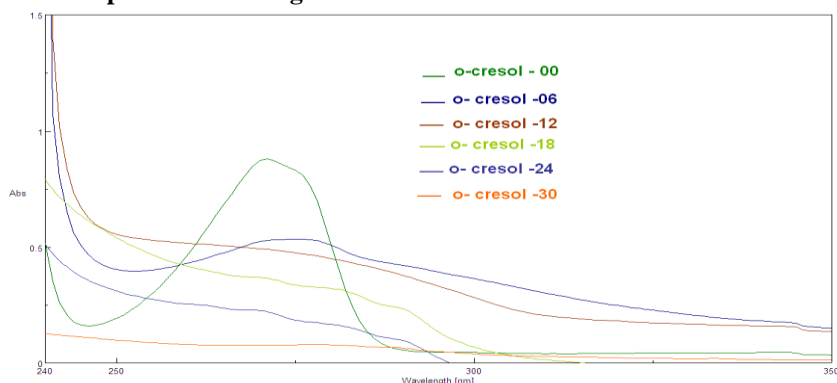


Fig.1 Spectrophotometric pattern of biodegradation by *Bacillus badius* D1

The (Fig. 1) showed that UV-Vis spectrum of o-cresol. Initially the spectrum was observed at 270 nm at 00 hr. Later on it changed to various levels suggested o-cresol has undergone several metabolic transformations.

2.4 Percentage degradation of cresol at various concentration :

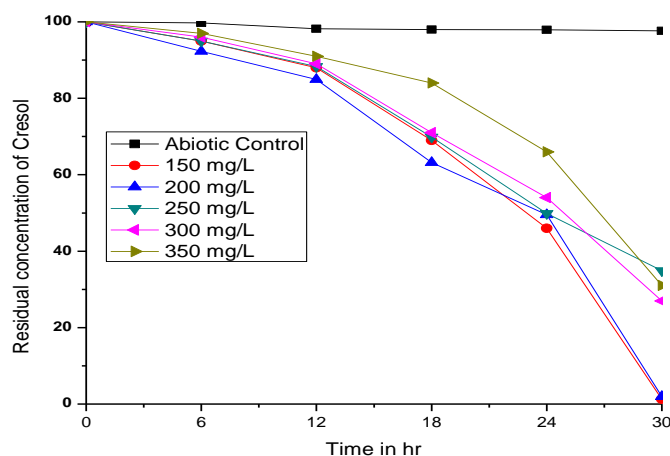
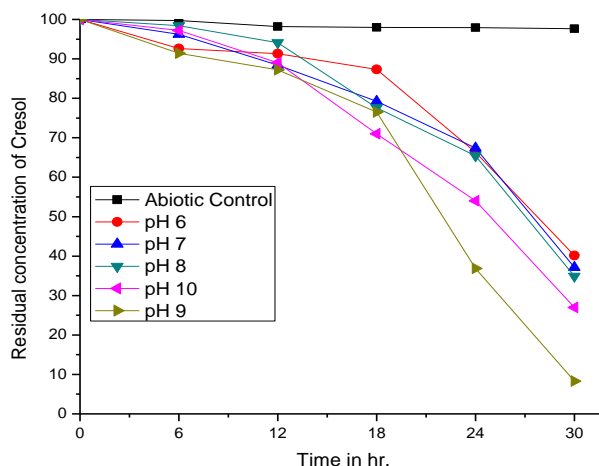


Fig.1.1 Percentage degradation of cresol at various concentration

As all microbes grow at pH 7.00 and 37 °C. The experiments were carried at the same parameters and the residual concentration of cresol at various time intervals has shown in (Fig 1.1). It indicates that various concentrations ranging from (150 mg/L -350 mg/L) was almost degraded by *Bacillus badius* D1 within 30 hr. At concentrations 150 mg/L and 200 mg/L no residual concentration observed after 30 hrs while at higher concentrations more than 75% degradation was observed.

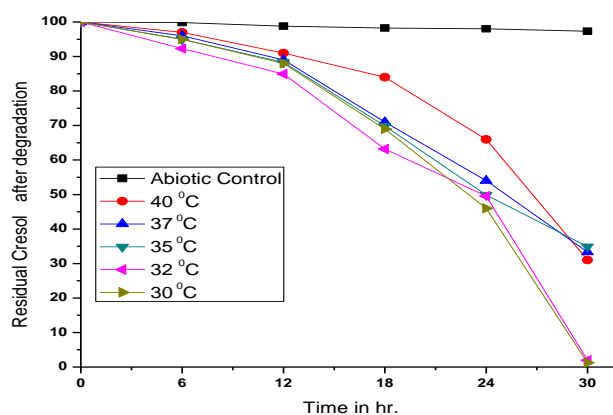
2.5 Percentage degradation of cresol at various pH:



Percentage degradation of cresol at various pH (Fig 1.2)

The (Fig. 1.2) is self explanatory. It showed that the pH 9.00 was more convenient for degradation as compared to other pH. The highest (350 mg/L) experimental concentration observed 92% degraded at pH 9.00.

2.6 Percentage degradation of cresol at various temperatures:



Percentage degradation of cresol at various temperatures (Fig 1.3)

The (Fig 1.3) suggested that the optimum temperature 32 °C was optimum. At this temperature and pH 9.00 the highest experimental concentration was completely degraded.

2.7 GCMS fragmentation data of Salicylic acid

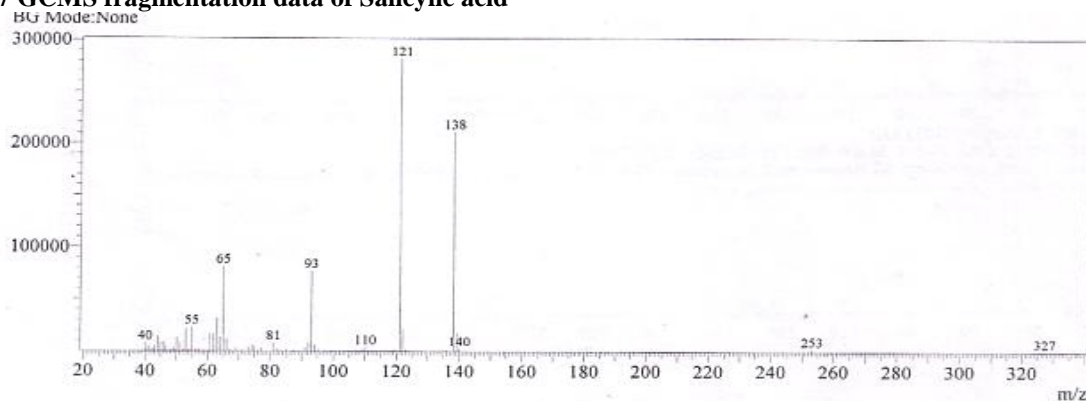


Fig.14 GCMS fragmentation pattern of Salicylic acid

The (Fig.1.4) showing fragmentation pattern was of salicylic acid had been confirmed by software GC Solution.

2.8 GCMS fragmentation data of catechol:

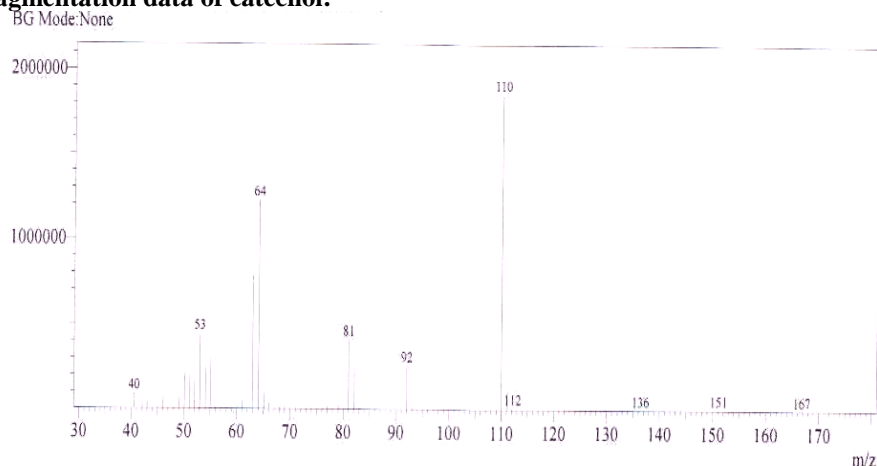


Fig.1.5 GCMS data of catechol as an intermediate metabolite

The (Fig.1.4 and 1.5) is showing fragmentation pattern had been confirmed by software GC Solution as Salicylic acid and Catechol respectively.

2.9 Enzyme induction Study:

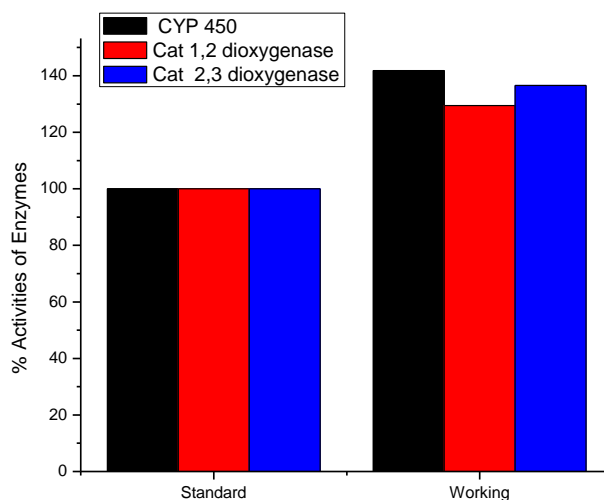


Fig. 1.6 Enzyme induction Study

CYP 450 and Catechol dioxygenase induction after addition of experimental concentration of cresol has shown in (Fig 1.6). It indicated the induction of CYP 450, Cat 1,2 dioxygenase and Cat 2,3 dioxygenase. Comparative to Cat 1,2 dioxygenase Cat 2,3 dioxygenase activity observed more.

III. Discussion:

Cresols are natural components of crude oil and coal tar. These are obtained by fractional distillation as well as synthesized in laboratory. During these processes cresols are released in atmosphere by natural and anthropogenic activities like manufacture, use, transport, and storage [12].

O-cresol has been used in disinfectants, fumigants in dyes, odors, in photographic film developers, pesticides etc [13]. It is carcinogenic and affects the kidney; lungs and also the nervous system [14]. Most of all phenolic compounds including cresols are harmful to both humans and our environment. The compounds are highly stable and soluble in water hence, create hazards to the environment [15]. Although other

photocatalytic or physicochemical methods are available to remove the pollutants from water bodies but these methods are harsh and costly. Therefore biodegradation is very good alternative to these methods.

There are several reports regarding the bio-degradation of cresols by neutrophilic organisms like bacteria and fungi but very scarce or no reports found on alkaliphilic biodegradation of cresol. Therefore an attempt was made to investigate the percent degradation of o- cresol by alkaliphilic strain *Bacillus badius* D1. Several changes in o-cresol original spectrum showed that it has undergone different bio-transformations (**Fig.1**)

The degradation of aromatic compounds by mesophilic micro-organisms and the oxidative breakdown of phenolic compounds have been studied in case of aerobic and anaerobic bacteria as well as yeasts [16]. In case of anaerobic biodegradation of o-, p-, and m- cresols; p-cresol was preferentially degraded while m-isomer and o- isomer remain persistent [17, 18]. In case of *Alkaligen eutrophus* phenol and cresol reduced from higher concentration to lower from soil [19, 20]. Similarly our strain *Bacillus badius* D1 degraded high concentration of o-cresol. (**Fig 1.1**) in very less time. The study was also extended to other parameters like variable pH (**Fig.1.2**) and temperature (**Fig.1.3**). It made to know the optimum pH 9.00 and temperature 32 °C for biodegradation.

The attempt was also made to isolate few of the of metabolites by solvent extraction and GCMS. The metabolites identified from fragmentation pattern were salicylic acid and catechol (**Fig 1.4 and Fig 1.5**). Earlier report shown 4-Hydroxybenzylsuccinate in culture flasks of anaerobic *Desulfobacterium cetonicum* feeding on p-cresol [21]. The observations regarding enzyme studies showed the enzyme activities of CYP 450, cat 1,2 dioxygenase and Cat 2,3 dioxygenase indicated that *Bacillus badius*D1 actively involved in o-cresol degradation.(**Fig. 1.6**). CYP 450 and Cat 2,3 dioxygenase activity was more than Cat 1,2 dioxygenase in case of *Bacillus badius* D1. The earlier report of aromatic hydrocarbon degradation using archaeobacteria [22] noted the induction of enzymes catechol 1,2 dioxygenase, protocatechuate 3,4 dioxygenase, phenol hydroxylase, acetanilide hydroxylase etc [23, 24].

Although; other reports [25, 26, 27, 28, 29] were observed on biodegradation of cresols the Strain *Bacillus badius* D1 showed higher degradation of o-cresol in very less time. It is nonpathogenic alkaliphile. Photocatalytic and physic-chemical methods are being employed in detoxification of cresols parallelly using nano particles [30, 31, 32]. Few other strains like *Lysinibacillus cresolivorans* [33] *Staphylococcus aureus* [34] *Aspergillus fumigatus* [35] were exploited for biodegradation recently.

IV. Conclusion:

Bacillus badius D1 is nonpathogenic alkaliphile have high pollution remediation measure and it could be employed at bioremediation site as a potent strain.

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